

Desplazamiento de Wienn ;

Clear [h, k, c, x]

Solve [1 - 1/Exp[x] - x/3 == 0, x]

Solve::ifun : Inverse functions are being used by Solve, so some solutions may not be found; use Reduce for complete solution information. >>

Out[3]= $\left\{ \left\{ x \rightarrow 0 \right\}, \left\{ x \rightarrow 3 + \text{ProductLog} \left[-\frac{3}{e^3} \right] \right\} \right\}$

In[4]:= $N \left[3 + \text{ProductLog} \left[-\frac{3}{e^3} \right], 100 \right]$

Out[4]= 2.82143937212207889340319133029448519534588174407311409227985769394121430450551739124568656534783.9644

Cálculo completo en Frecuencia ;

In[6]:= $\rho[f_, T_] := 8 \pi / c^3 f^2 h f / (\text{Exp}[h f / (k T)] - 1)$

In[7]:= $D[\rho[f, T], f]$

Out[7]=
$$\frac{24 f^2 h \pi}{c^3 \left(-1 + e^{\frac{f h}{k T}} \right)} - \frac{8 e^{\frac{f h}{k T}} f^3 h^2 \pi}{c^3 \left(-1 + e^{\frac{f h}{k T}} \right)^2 k T}$$

In[8]:= $\text{Solve} \left[\frac{24 f^2 h \pi}{c^3 \left(-1 + e^{\frac{f h}{k T}} \right)} - \frac{8 e^{\frac{f h}{k T}} f^3 h^2 \pi}{c^3 \left(-1 + e^{\frac{f h}{k T}} \right)^2 k T} == 0, f \right]$

Solve::ifun : Inverse functions are being used by Solve, so some solutions may not be found; use Reduce for complete solution information. >>

Out[8]= $\left\{ \left\{ f \rightarrow \frac{k T \left(3 + \text{ProductLog} \left[-\frac{3}{e^3} \right] \right)}{h} \right\} \right\}$

In[9]:= $x = 3 + \text{ProductLog} \left[-\frac{3}{e^3} \right];$

In[10]:= << PhysicalConstants`

In[11]:= $h = \text{PlanckConstant}$

Out[11]= 6.62607×10^{-34} Joule Second

In[12]:= $k = \text{BoltzmannConstant}$

Out[12]= $\frac{1.38065 \times 10^{-23} \text{ Joule}}{\text{Kelvin}}$

In[13]:= $c = \text{SpeedOfLight}$

Out[13]= $\frac{299792458 \text{ Meter}}{\text{Second}}$

In[14]:= $f_{\text{max}} \text{toT} = k / h x$

Out[14]= $\frac{5.87893 \times 10^{10}}{\text{Kelvin Second}}$

Cálculo en Longitud de Onda ;

Clear [h, k, c, x]

In[17]:= $\rho\lambda[\lambda_, T_] := 8 \pi h c / (\lambda^5) 1 / (\text{Exp}[h c / (\lambda k T)] - 1)$

In[18]:= $D[\rho\lambda[\lambda, T], \lambda]$

$$\text{Out[18]} = \frac{8 c^2 e^{\frac{c h}{k T \lambda}} h^2 \pi}{\left(-1 + e^{\frac{c h}{k T \lambda}}\right)^2 k T \lambda^7} - \frac{40 c h \pi}{\left(-1 + e^{\frac{c h}{k T \lambda}}\right) \lambda^6}$$

In[19]:= $\text{NSolve}\left[\frac{8 c^2 e^{\frac{c h}{k T \lambda}} h^2 \pi}{\left(-1 + e^{\frac{c h}{k T \lambda}}\right)^2 k T \lambda^7} - \frac{40 c h \pi}{\left(-1 + e^{\frac{c h}{k T \lambda}}\right) \lambda^6} == 0, \lambda\right]$

NSolve::nsmet : This system cannot be solved with the methods available to NSolve. >>

In[20]:= $\text{Solve}[1 - 1/\text{Exp}[x] - x/5 == 0, x]$

Solve::ifun : Inverse functions are being used by Solve, so some solutions may not be found; use Reduce for complete solution information. >>

Out[20]= $\left\{\{x \rightarrow 0\}, \left\{x \rightarrow 5 + \text{ProductLog}\left[-\frac{5}{e^5}\right]\right\}\right\}$

In[21]:= $\text{N}\left[5 + \text{ProductLog}\left[-\frac{5}{e^5}\right], 10\right]$

Out[21]= 4.965114232

In[22]:= $x = 5 + \text{ProductLog}\left[-\frac{5}{e^5}\right];$

In[35]:= $h = \text{PlanckConstant};$
 $k = \text{BoltzmannConstant};$
 $c = \text{SpeedOfLight};$
 $\hbar = \text{PlanckConstantReduced};$

In[26]:= $\lambda_{\text{maxT}} = h c / (k x)$

Out[26]= 0.00289777 Kelvin Meter

Test λ_{max} fmax;

Out[27]= fmax Test λ_{max}

In[28]:= $f_{\text{maxtoT}} \lambda_{\text{maxT}}$

Out[28]= $\frac{1.70358 \times 10^8 \text{ Meter}}{\text{Second}}$

In[29]:= $h c$

Out[29]= $1.98645 \times 10^{-25} \text{ Joule Meter}$

In[33]:= $\text{Convert}[h c, \text{ElectronVolt Angstrom}]$

Out[33]= 12398.4 Angstrom ElectronVolt

In[39]:= $\text{Convert}[\hbar c, \text{ElectronVolt Angstrom}]$

Out[39]= 1973.27 Angstrom ElectronVolt

12398.418754199978` Angstrom ElectronVolt

1973.2696312541852` Angstrom ElectronVolt

Stefan – Boltzmann,

In[43]:= $\text{Clear}[h, k, c, x]$

In[44]:= $\rho[f_, T_] := 8 \pi / c^3 f^2 h f / (\text{Exp}[h f / (k T)] - 1)$

In[45]:= **Integrate** [$\rho[f, T]$, { f , 0, Infinity}]

Out[45]= ConditionalExpression $\left[\frac{8 k^4 \pi^5 T^4}{15 c^3 h^3}, \text{Re} \left[\frac{h}{k T} \right] > 0 \right]$

In[46]:= **h** = PlanckConstant;
k = BoltzmannConstant;
c = SpeedOfLight;
ħ = PlanckConstantReduced;

In[50]:= **b** = $\frac{8 k^4 \pi^5}{15 c^3 h^3}$

Out[50]= $\frac{7.56577 \times 10^{-16} \text{ Joule}}{\text{Kelvin}^4 \text{ Meter}^3}$

In[51]:= **Convert** [**b**, ElectronVolt / (Kelvin ^ 4 Meter ^ 3)]

Convert::temp : Warning: Convert[old,new] converts units of temperature. ConvertTemperature[temp,old,new] converts absolute temperature. >>

Out[51]= $\frac{4722.18 \text{ ElectronVolt}}{\text{Kelvin}^4 \text{ Meter}^3}$

Nro Fotonos ;

In[52]:= **Clear** [**h**, **k**, **c**, **x**]

In[53]:= **Nf** [**f**_, **T**_, **V**_] := $8 \pi / c^3 f^2 h f / (\text{Exp}[h f / (k T)] - 1) V / (h f)$

In[54]:= **Integrate** [**Nf** [**f**, **T**, **V**], {**f**, 0, Infinity}]

Out[54]= ConditionalExpression $\left[\frac{16 k^3 \pi T^3 V \text{Zeta}[3]}{c^3 h^3}, \text{Re} \left[\frac{h}{k T} \right] > 0 \right]$

In[55]:= **N** [**Zeta** [3], 10]

Out[55]= 1.202056903

In[56]:= **h** = PlanckConstant;
k = BoltzmannConstant;
c = SpeedOfLight;
ħ = PlanckConstantReduced;

In[60]:= $\frac{16 k^3 \pi \text{Zeta}[3]}{c^3 h^3}$

Out[60]= $\frac{2.02869 \times 10^7}{\text{Kelvin}^3 \text{ Meter}^3}$

In[62]:= $2.0286918677723482 \cdot 10^7 \cdot 298^3 \cdot (0.01)^3$

Out[62]= 5.36865×10^8

In[63]:= **b** = $\frac{8 k^4 \pi^5}{15 c^3 h^3}$

Out[63]= $\frac{7.56577 \times 10^{-16} \text{ Joule}}{\text{Kelvin}^4 \text{ Meter}^3}$

In[65]:= $(b T^4) / \left(\frac{16 k^3 \pi T^3 \text{Zeta}[3]}{c^3 h^3} \right)$

Out[65]= $\frac{3.72938 \times 10^{-23} \text{ Joule T}}{\text{Kelvin}}$

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In[66]:= Convert [3.729382531620919`*^-23 Joule , ElectronVolt]
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Out[66]= 0.00023277 ElectronVolt
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In[67]:= 0.0002327697704891433` ElectronVolt 298
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Out[67]= 0.0693654 ElectronVolt
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In[70]:= 12398 / (0.0002327697704891433 × 5800)
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Out[70]= 9183.26
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